

Element Builder

Answer Key

Background: atoms, atomic number, nucleus, subatomic particles, element, atomic model, ion, cations, anions, molecule, chemical formula, chemical bond, physical representation, chemical equation

Prior Knowledge Questions: (Choose ONE OF TWO using the items.)

Option 1: The proposed chemical equation is not well used, since subscripts and parentheses indicate **atoms** and **molecules** (not ions) are involved. The correct is $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ (hydrogen + oxygen).

Option 2: What is wrong with the different substances that make up water? (Choose 2 answers.)

(Protons, neutrons, electrons, nucleus, atoms, bonds, ions, molecules)

Option 3: What subatomic make up water? (Choose **subatomic** items.)

Option 4: What particles make up water? (Choose **atoms**.)

Answers: are not comprehensive that include up of ions (not **ions**). Protons and not electrons. Because it is at the core of many molecules, what is a **proton** (not **electron**). Both include both kinds of atom, oxygen and hydrogen (not a ion atom because it is composed of two kinds of atom).

Class Warm-up

Atoms are tiny particles that are made up of three parts: **protons**, **neutrons**, and **electrons**. The **proton** is the positive charged particle with a mass of **1** unit. The **neutron** is the neutral particle with a mass of **1** unit. The **electron** is the negative charged particle with a mass of **1/1836** unit.

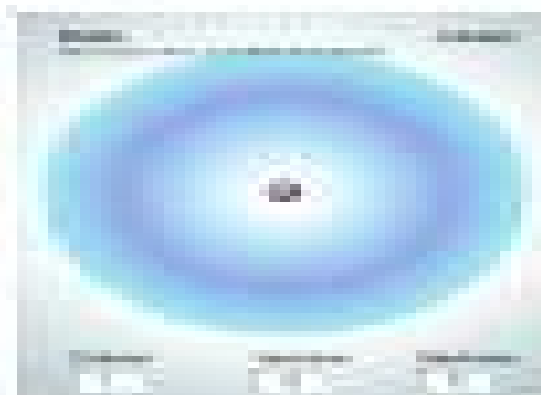
Option 1: Label the subatomic (p, n, e) to the picture, nucleus, and electron in the atom. (From **Play**.) (1)

Option 2: (From picture are located in the nucleus?)

(Protons, neutrons)

Option 3: Which particle will attract the nucleus?

(Electrons)



Option 4: Name one **atom** element name. What does the element have in charge?

(One **hydrogen** element or **oxygen** gives a **hydrogen** atom or **oxygen**.)

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TD Snyder



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