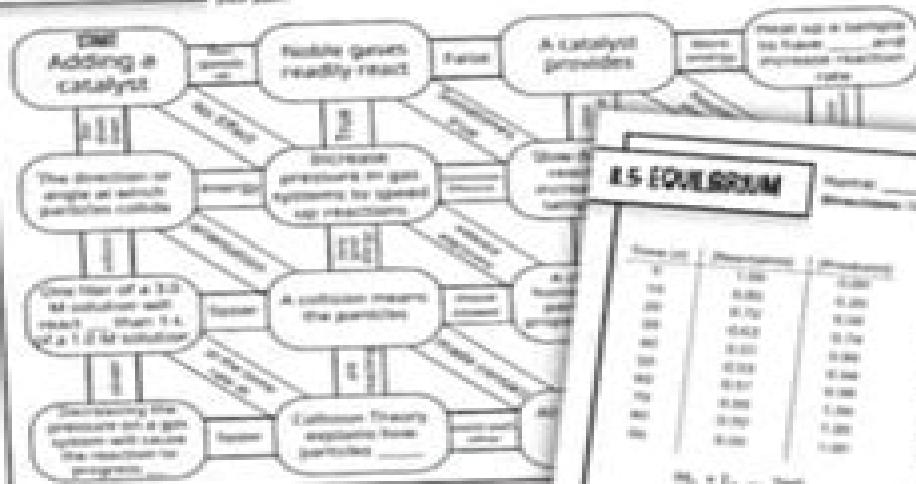


KINETICS & EQUILIBRIUM QUICK PRACTICE SHEETS

8.1 COLLISION THEORY



8.5 EQUILIBRIUM

Time (s)	Concentration of H_2 (mol dm ⁻³)	Concentration of I_2 (mol dm ⁻³)
0	1.000	1.000
10	0.860	0.860
20	0.720	0.720
30	0.640	0.640
40	0.580	0.580
50	0.530	0.530
60	0.500	0.500
70	0.480	0.480
80	0.460	0.460
90	0.450	0.450
100	0.440	0.440



8.6 LE CHATELIER'S PRINCIPLES



1. Add ammonium chloride
2. Remove ammonium chloride
3. Add ammonia
4. Add hydrochloric acid
5. Remove ammonia
6. Remove hydrochloric acid
7. Increase temperature
8. Add a noble gas
9. Decrease temperature
10. Add a catalyst

6 WORKSHEETS



QUICK & EFFECTIVE

Kinetics And Equilibrium Mark Rosengarten Homework Answer

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Kinetics And Equilibrium Mark Rosengarten Homework Answer Book Review: Unveiling the Power of Words

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