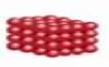


# Kinetic Molecular Theory

#### I. MOLECULES IN MOTION:

- A.Kinetic Molecular Theory (KMT) = the idea that particles of matter are always in motion and that this motion has consequences.
  - 1) theory developed in the late 19th century to account for the behavior of the atoms and molecules that make up matter
  - 2) based on the idea that particles in all forms of matter are always in motion and that this motion has consequences
  - 3) can be used to explain the properties of solids, liquids, and gases in terms of the energy of particles and the forces that act between them
  - 3 assumptions of KMT:
    - All <u>matter</u> (solid, liquid, gas) is made of <u>particles</u> (atoms, ions, molecules)
    - 2.Particles are in constant motion (electrons move)
      - · Particles of a GAS travel in completely random motion
      - Particles of a <u>LIQUID</u> appear to vibrate around moving points
      - Particles of a <u>SOLID</u> appear to vibrate around fixed points







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